





# Thermal energy storage implementation using phase change materials in a solar cooling and refrigeration applications

International Conference on Solar Heating and Cooling for Buildings and Industry San Francisco (USA)

July 9- 11, 2012



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#### Introduction

- Nowadays, solar cooling and refrigeration technology have become vital for both human comfort since solar energy is the cheapest and most extensively available renewable energy
- In solar vapor absorption systems, the energy received from the solar collector is given as heat input to the generator; hence it has to assure a constant heat input to the absorption chiller during all the process

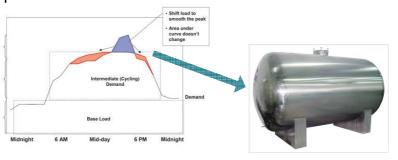


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## Introduction

- Here the main disadvantage of this energy resource is the mismatch between the energy supply and the energy demand
- Therefore, when energy is available but cannot be given to the process, thermal energy storage (TES) may become an important issue



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# Objective

- To select and to test different phase change materials (PCM) candidates in lab scale
- To compare the PCM selected at pilot plant scale in a real high temperature system

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# Materials and methodology

- Selection of the PCM
  - Storage temperature 140 − 200 °C
  - Heat of fusion > 150 kJ/kg

Material	Experimental phase change temperature [°C]	Experimental phase change enthalpy [kJ/kg]	
Salicylic acid	159.1 (m) / 111.3 (s)	161.5 (m) / 109.4 (s)	
Benzanilide	163.6 (m) / 136.1 (s)	138.9 (m) / 129.4 (s)	
D-mannitol	166.8 (m) / 117 (s)	260.8 (m) / 214.4 (s)	
Hydroquinone	172.5 (m) / 159.5 (s)	235.2 (m) / 178.7 (s)	
Potassium thiocynate	176.6 (m) / 156.9 (s)	114.4 (m) / 112.5 (s)	

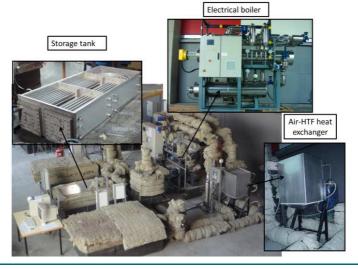
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## Materials and methodology

• High temperature pilot plant

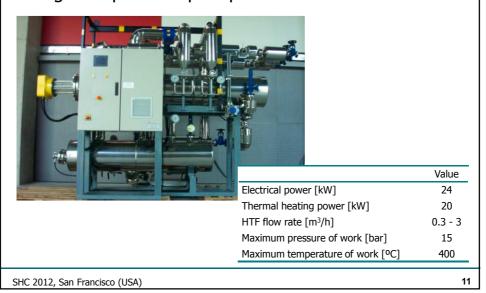


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# Materials and methodology

• High temperature pilot plant - boiler





## Materials and methodology

High temperature pilot plant – air heat exchanger



	Value	
Thermal cooling power [kW]	20	
Air flow rate [m³/h]	1800	
Air inlet temperature [°C]	ambient	
Exterior dimensions [mm]	700x540x440	

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# Materials and methodology

• High temperature pilot plant - tank



Storage tank width [m] 0.527
Storage tank height [m] 0.273
Storage tank depth [m] 1.273
Number of HTF tubes [-] 49
HTF pipes average length [m] 2.90
Heat transfer surface [m²] 6.568
Total d-mannitol mass [kg] 160

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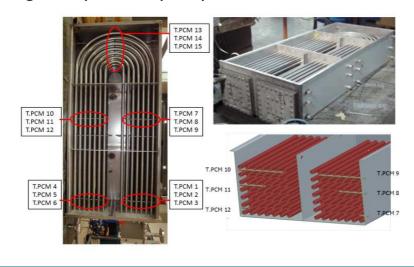
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Value



## Materials and methodology

• High temperature pilot plant - tank



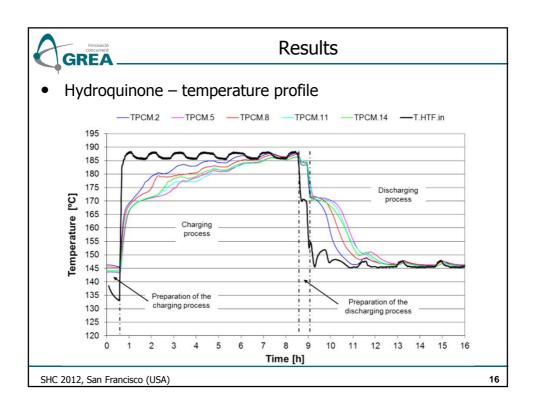
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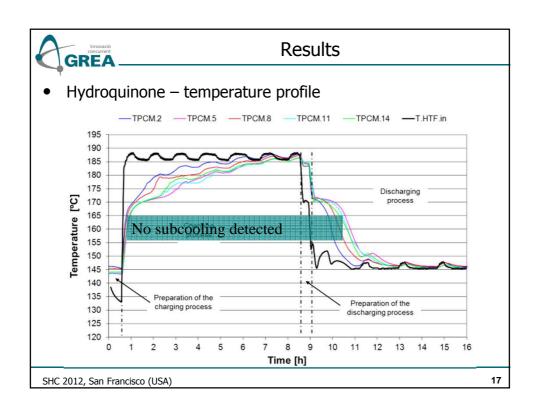


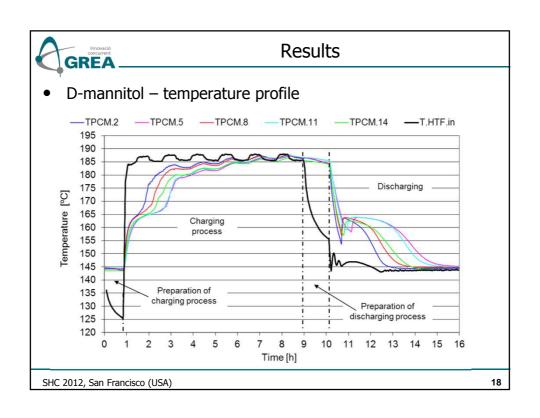
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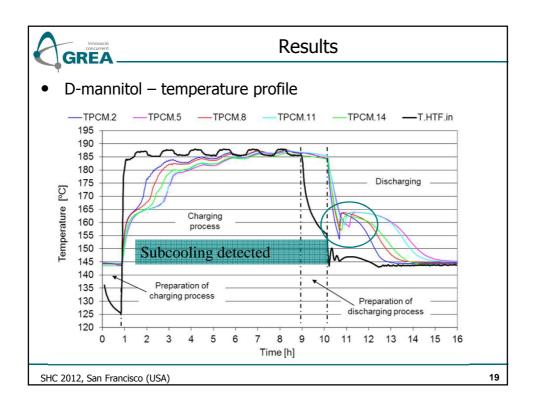
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## Results

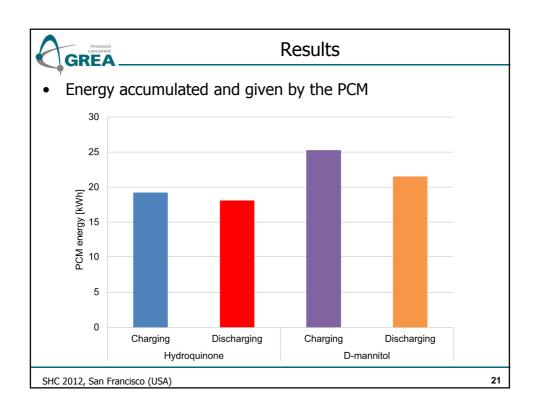
• Energy balance in the storage tank

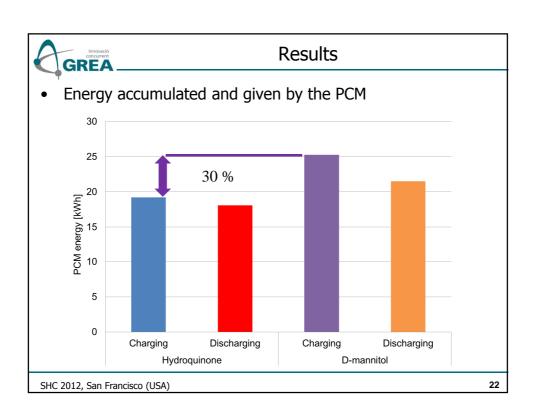
$$-\ Q_{HTF} = Q_{PCM} + Q_{tank} + Q_{loss} + Q_{acc\,HTF} + Q_{insulation}$$

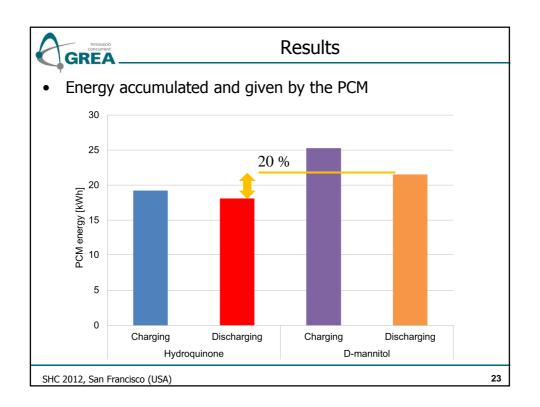
$$- \ \, \varepsilon_{\rm cha} = \frac{\rm Q_{\rm PCM}}{\rm Q_{\rm HTF}} \qquad \qquad \varepsilon_{\rm dis} = \frac{\rm Q_{\rm HTF}}{\rm Q_{\rm PCM} + Q_{\rm steel}} \label{eq:epsilon}$$

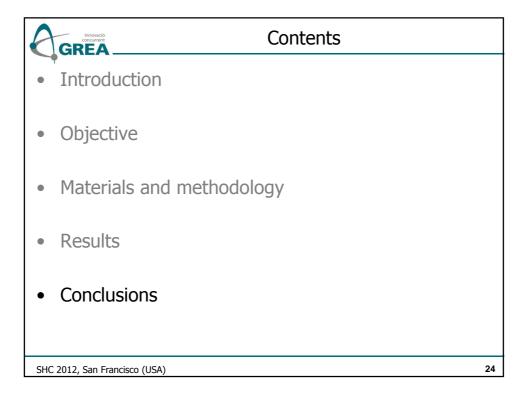
	Hydroquinone		D-mannitol	
	Charging	Discharging	Charging	Discharging
Time [min]	476	362	486	354
$\Delta E_{HTF}$ [kWh]	21.5	17.9	27.7	21.4
$\Delta E_{PCM}$ [kWh]	19.2	18.1	25.3	21.5
$\Delta E_{loss}$ [kWh]	1.4	1.2	1.3	0.8
$\Delta E_{steel}$ [kWh]	0.8	0.5	0.9	0.5
ε[-]	0.9	0.96	0.91	0.97

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#### Conclusions

- A high temperature pilot plant with TES system based on shell-and-tubes heat exchanger were designed and built at the University of Lleida
- Literature and DSC research of some PCM candidates for solar cooling applications were done
- Hydroquinone and d-mannitol were the PCM selected to test them at pilot plant scale
- Different charging and discharging experiments with different flow rates and HTF temperatures were performed

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## Conclusions

- For both PCM, no hysteresis was detected, and even though hydroquinone presented subcooling in the DSC, it did almost not appear in pilot plant scale, however, when d-mannitol was used big subcooling was detected
- For the same boundary conditions, the energy stored by dmannitol was higher than that for hydroquinone
- The enhancement was about 30% and 20% during the charging and the discharging processes even though the enhancement of the latent heat was only 10 and 16%, respectively

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## Acknowledgements

- The work is partially funded by the Spanish government (ENE2011-22722). The authors would like to thank the Catalan Government for the quality accreditation given to their research group GREA (2009 SGR 534). Antoni Gil would like to thank the Col·legi d'Enginyers Industrials de Catalunya for his research appointment. Eduard Oró would like to thank the University of Lleida for his research fellowship.
- To all co-authors of this work



## Thank you for your atention

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